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THEORETICAL INVESTIGATIONS	•	SYNTHESIS OF	ORGANIC MAG	SNETIC POLY	MERS
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19. ABSTRACT (Continue on reverse if necessare Progress in theoretical and synthe			tic species is o	utlined by slid	des

Progress in theoretical and synthetic investigations of organic magnetic species is outlined by slides presented at the Office of Naval Research contractors' meeting of Fall, 1989. Overall, progress was made in synthesis of appropriate groups to allow photochemical and thermal generation of phenoxyl radicals in the solid phase. Thermal stability of a variety of substituted phenoxyl systems generated by these methods in PPMA matrices was investigated.

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Technical Report #10

THEORETICAL INVESTIGATIONS AND ATTEMPTED SYNTHESIS OF ORGANIC MAGNETIC POLYMERS AND OLIGOMERS

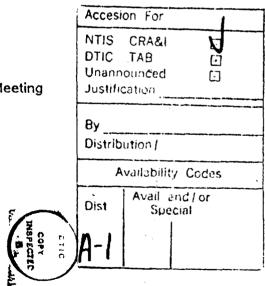
by Paul M. Lahti

Presented at

Office of Naval Research Contractors' Meeting Autumn, 1989

> University of Massachusetts Department of Chemistry Amherst, MA 01003

Submitted 6 March 1990



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THEORETICAL INVESTIGATIONS AND ATTEMPTED SYNTHESIS OF ORGANIC MAGNETIC POLYMERS AND OLIGOMERS

by Paul M. Lahti
Department of Chemistry
University of Massachusetts
Amherst, MA 01003

OBJECTIVE:

To investigate the possibilies of generating magnetic information storage materials from basically non-metallic organic precursors, in order to combine the technological usefulness of magnetism with the practical usefulness of organic polymeric materials.

APPROACH:

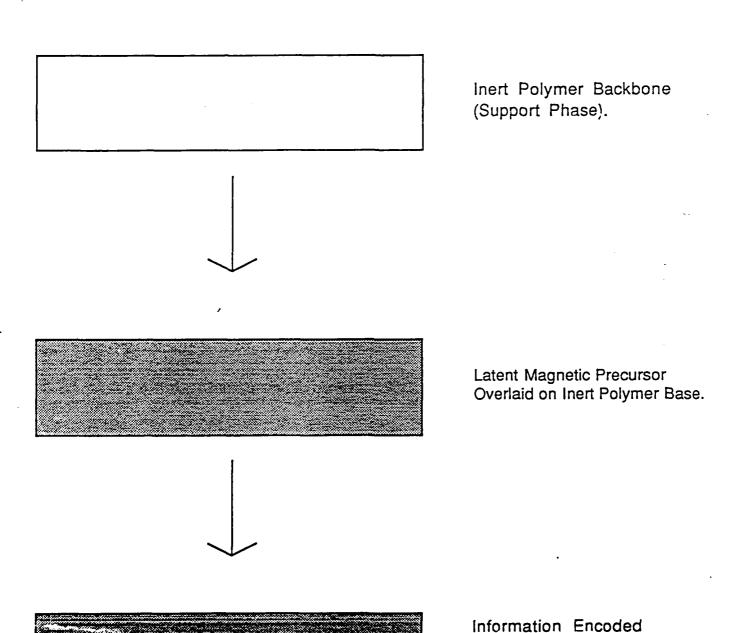
Theoretical computations allow selection of polyradical polymer structures likely to yield high-spin magnetic properties.

Experimentally we synthesize as model compounds small segments of a potential highly magnetic polymer, then test the predictions of theory by standard methods for investigating magnetic properties (electron spin resonance, magnetic susceptibility). Successful predictions for small models will imply that it is worth attempting synthesis of larger polymers predicted to have desirable magnetic properties.

RESULT:

Theory does predict that linkage of several radical units into an oligomer will in many cases yield high-spin organic polyradicals. Experimental generation of such species will constitute the first step toward creating truly polymeric magnetically imaged organic materials.

IDEALIZED USE OF POLYRADICAL MAGNETIC MATERIAL PERMANENT READ-ONLY MEMORY



By "Developing" Some Portion of Precursor To Become Magnetic.

INTEREST IN ORGANIC FERROMAGNETIC MATERIALS

J. S. Miller, A. J. Epstein, W. M. Reiff

Chem. Rev. 88, 201(1988), Accts. Chem. Res., 21, 114(1988).

R. Breslow, et al.

Pure Appl. Chem., **54**, 927(1982), J. Am. Chem. Soc., **109**, 6412(1987), J. Am. Chem. Soc., **110**, 3970(1988)

Yu. V. Korshak, T. V. Medvedeva, A. A. Ovchinnikov, V. N. Spector Nature, 126, 370(1987)

J. B. Torrance, S. Oostra, A. Nazzal

Synth. Metals, 19, 709(1987).

Experimental Work Toward Synthesis of High Spin Polyradicals

Examples include:

Cf. T. Sugawara, S. Bandow, K. Kimura, H. Iwamura, K. Itoh, J. Am. Chem. Soc. 1984, 106, 6449; 1986, 108, 368 and references therein.

H

$$n=3-5$$

hv

 $n=3-5$
 $n=3-5$
 $n=3-5$

high spin state polycarbenes observed, characterized

H. Nishide, N. Yoshioka, K. Inagaki, E. Tsuchida, Macromolecules, 1988, 21, 3120.

$$\begin{array}{c} OH \\ IBU \\$$

Stable, with about 1 free spin / 10 monomers by ESR spin counting.

Y. Miura, T. Ohana, J. Org. Chem., 1988, 53, 5770.

oligomers

DESIRED: A unimolecular strategy to generate phenoxyl radicals in solution OR in solid state.

STRATEGY

OUR PRESENT METHOD

ArO-C-C-O-O-
t
Bu \longrightarrow ArO \cdot CO₂ CO t BuO \cdot heat >70°C

ADVANTAGES

- 1. Easy and cheap to make
- 2. Fairly easy to store for some time, soluble in common solvents
- 3. Some photolability, easy thermal decomposition at > 70°C.

POSSIBLE FUTURE APPROACHES

O O
$$\parallel \parallel \parallel$$
 ArO-C-C-X $X = N=NR$, I, CPh₃

Synthesis of Peroxyoxalate Precursors for Phenoxy Systems

CMe₃

$$CHe_3$$

$$CHe_3$$

$$CHe_3$$

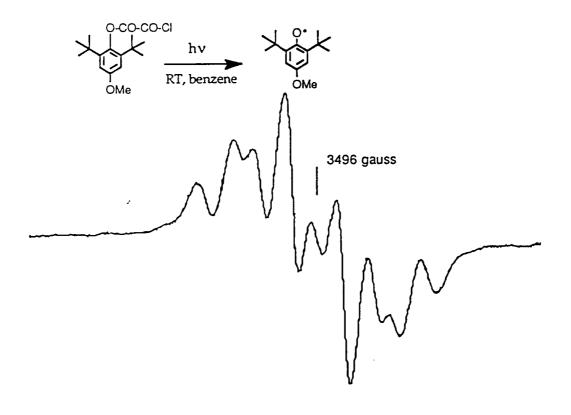
$$CMe_3$$

stable for several days at 25°C in benzene

ESR

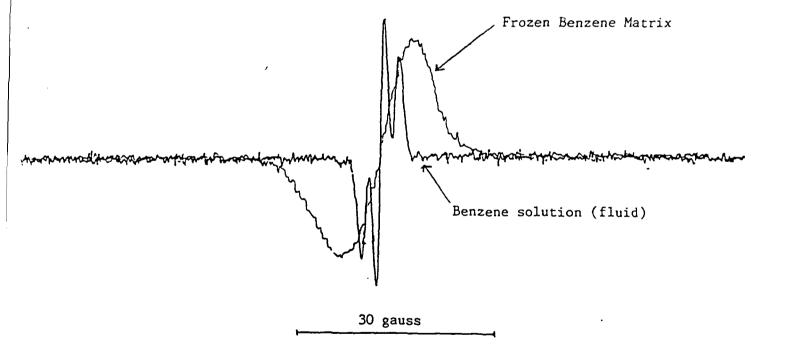
pentane recryst. at -78°C good room temperature stability

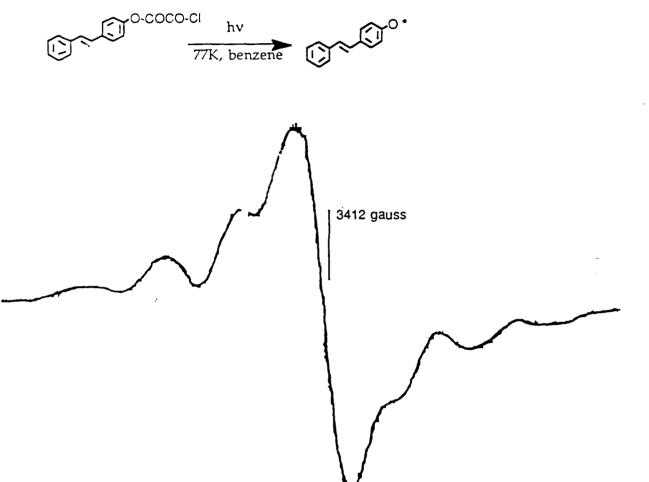
¹HNMR, IR



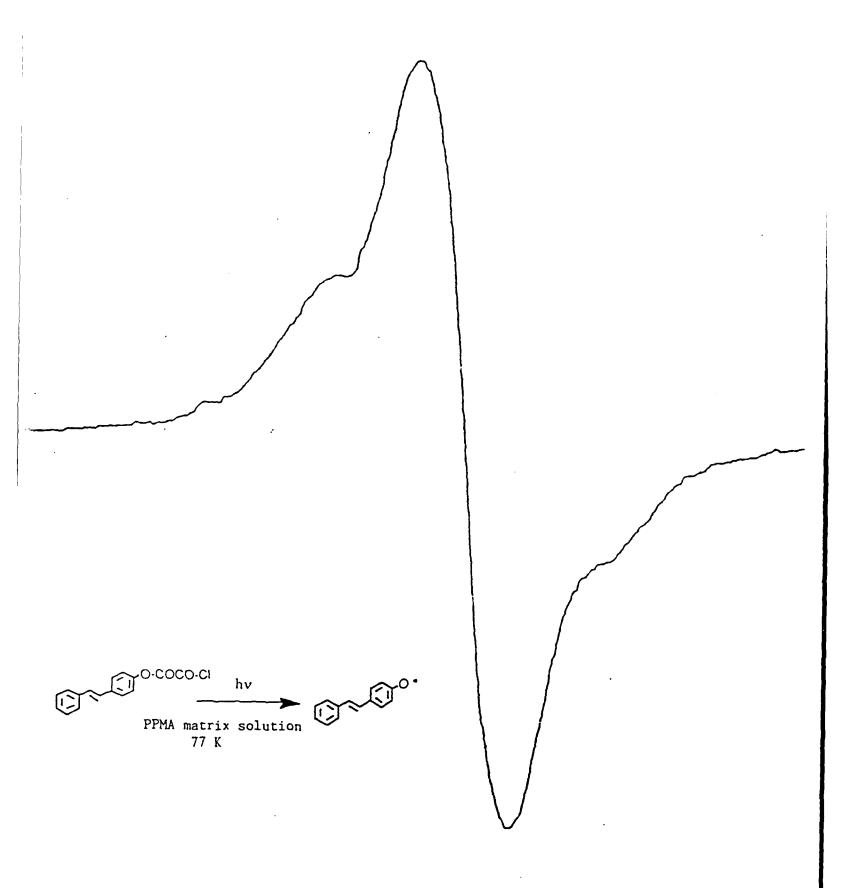
7.1 gauss

ESR Spectra in rigid matrix are much broader than in solution

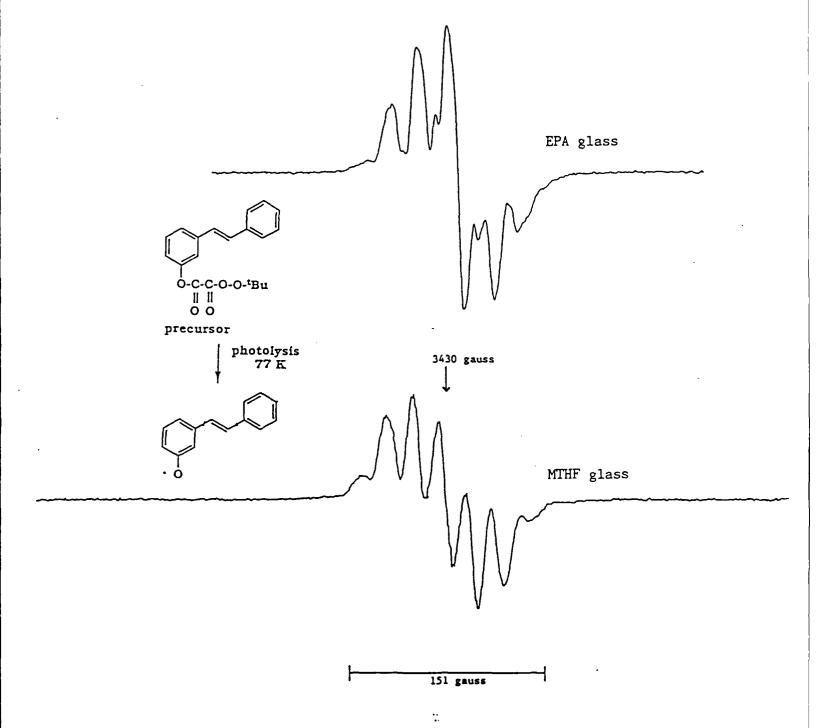




81 gauss



(SIMILAR LOSS OF FINE STRUCTURE IS OBSERVED IN NEAT SAMPLE IRRADIATION)



Synthesis of Dioxy-Stilbenes

meta, meta' meta, para'

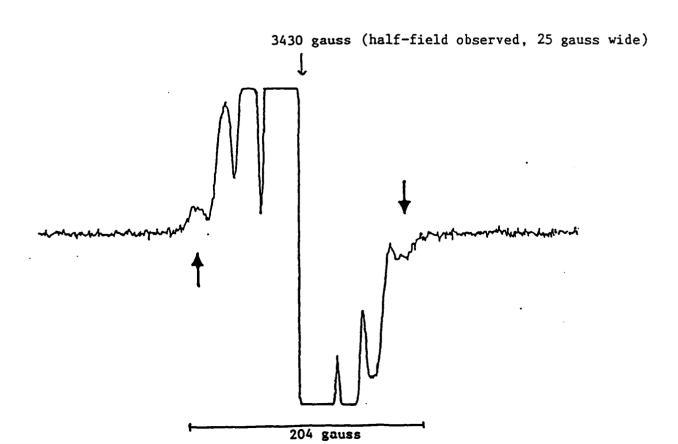
(X = -C1)

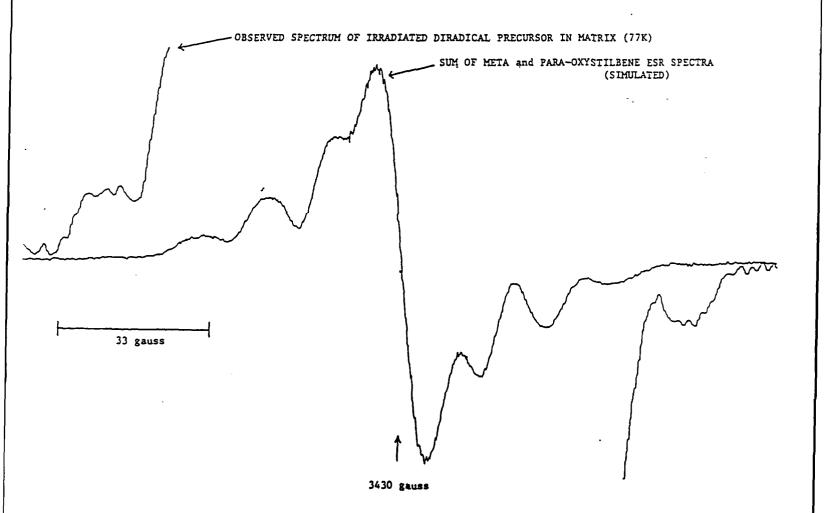
Ι

precursor

Theory predicts a paramagnetic triplet ground state for this molecule.

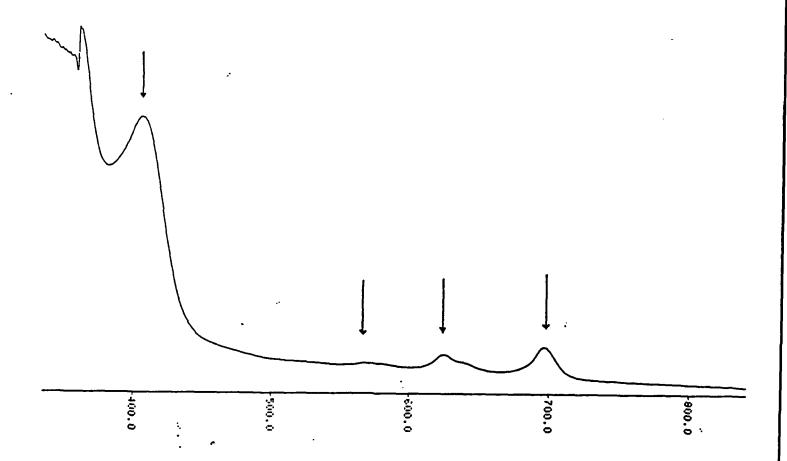
> diradical is produced when both centers are generated





STABILITY OF UNHINDERED PHENOXYS IS EASILY MONITORED IN POLYMERIC MATRIX BY UV-vis SPECTROSCOPY.

 $AOB = -O-COCO-OO^{\dagger}Bu$



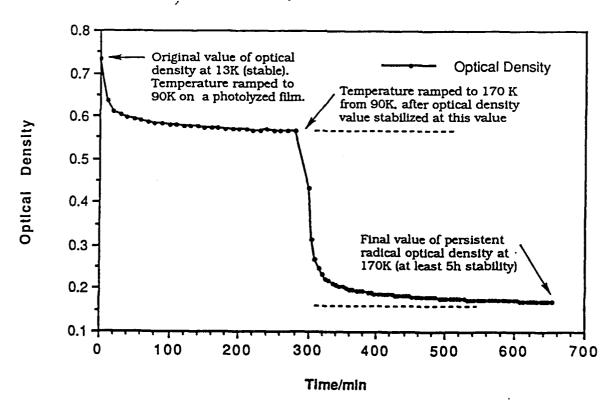
(Arrows indicate the para-stilbenoxy absorbances)

UNHINDERED RADICAL STABILITY IN MATRIX

Unhindered para-stilbenoxy appears indefinitely stable by UV-vis when generated in PPMA at 13K. When warmed swiftly (<4 min) to 90K, rapid absorbance decay occurs, which slows and leaves the absorbance (radical concentration) approaching an asymptotic value. Warming again to 170K results in the same behavior, with the final result being a much decreased (relative to the 13K value) but still finite amount of radical with stability for at least 5h at 170K. At room temperature, the radical appears to be completely quenched. Similar behavior occurs in polystyrene, but onset of decay behavior is faster. Some decay in both matrices has already occured by 77K with our typical 10% wt/wt radical-in-polymer mixes.

With some optimization, further increase in stabilization of unhindered radicals in polymer matrix may be possible.

DECAY OF p-STILBENOXY IN PPMA



A NEW, EFFICIENT SYNTHESIS OF para-OLEFINIC 2,6-di-tert-Butylphenolic polymer precursors.

(previously reported in D. Braun, B. Meier, *Makromol Chem.*, 167, **119**(1973))

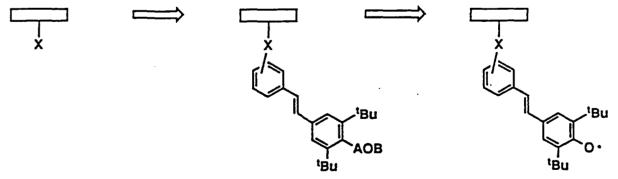
further functionalization of this polymer to the polyradical is desirable.

FUTURE WORK

1. Nitrenes and polynitrenes

2. Poly(2,6-di-*tert*-butyl-4-oxy-styrene) derivatives and copolymers. Polymers with pendant hindered stilbenoxy groups.

 t BuOO(CO)(CO)OO t Bu



highly paramagnetic?

3. Optimize radical production in polymer matrices

ACKNOWLEDGEMENTS

Group Members

David Modarelli Frank Rossitto Masaki Minato Andrew Ichimura Frank R. Denton III Ananda Sarker Mark Kearley Organic Magnetic Materials
Organic Magnetic Materials
Organic Magnetic Materials
Computational (Polyradicals)
Conducting Polymers
Conducting Polymers
Pentamethylenepropanes

Support

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